SOV/89-6-3-12/29

An Apparatus for Removing Oxygen and Water Vapor Traces From Inert Gases

gel. The necessary vacuum is produced with a forepump. Before the beginning of work all volumes are evacuated, then filled with inert gas and again evacuated. The containers with the absorbers are then heated to 200 - 250°C (silica gel) and 150 - 200°C (Na+K). Through a number of valves the gas comes first into a stock bin, then into a pressure-reducing valve and finally into the containers filled with the absorbers. The passing amount of gas is measured by means of a rotation flux meter. The purified gas is stored in a gas bottle. The regeneration of the consumed (oxidized) coppered silica gel is carried out by means of hydrogen stored in a steel bottle which passes a pressure-reducing valve and a rotation meter and then enters the steel container 1. Hydrogen consumption is in this case about 40 l per hour. After the first cylindrical container is filled with hydrogen during 10 minutes the heating coil of this device is switched on. At 250 - 300°C an intense reduction of copper oxide starts. The regeneration process must start when approximately 200  $m^3$  gas are purified. It takes approximately 8 hours. For the determination of small amounts of oxygen in the gas the well-known colorimetric method was used (color

Card 2/3

Traces SOV/89-6-3-12/29
An Apparatus for Removing Oxygen and Water Vapor/From Inert Gases

change of a copper solution containing ammonia). About 15 - 20 minutes are necessary for determining the oxygen content. In this case an accuracy of 10<sup>-5</sup>% can be attained. The different working conditions to obtain maximum efficiency were investigated and the observed dependence of gas purity on the rate of flow of the gas and on the temperature of the absorber is graphically represented. From this it may be seen e.g. that at a consumption of 60 l argon/h the oxygen content is 0.0003 per cent by volume in the purified gas. The purification of nitrogen at the same amount of flow reduces the oxygen content of nitrogen to 0.003 per cent by volume. There are 3 figures and 2 Soviet references.

SUBMITTED:

June 21, 1958

Card 3/3

83142 8/170/60/003/006/005/011 BO13/BO67

11.4140 AUTHORS:

TITLE:

Grachev, N. S., Kirillov, P. L.

Experimental Determination of the Elasticity of Potassium

Vapors at Temperatures of 550 - 1280°C

Inzhenerno-fizicheskiy zhurnal, 1960, Vol. 3, No. 6, PERIODICAL:

pp. 62 - 65

TEXT: The apparatus schematically shown in Fig. 1 was used to determine the elasticity of potassium vapors between 550 and 1280°C. The vapor pressure was measured by means of a compensating manometer with a sensitivity of 1 mm torr. The results of experimental series with 203 measurements are shown in a diagram (Fig. 2). All experiments were in good agreement within the accuracy of measurement. For comparison, data from Ref. 2 are also plotted. The results obtained by the authors are lower than those of Ref. 2. At 1280°C the divergence is 20%. This divergence could not be explained. The equation log  $P_g = -\frac{4970}{T} - 0.5 \log T + 6.160$ 

Card 1/2

83142

Experimental Determination of the Elasticity of Potassium Vapors at Temperatures of 550-1280°C

S/170/60/003/006/005/011 B013/B067

gives an approximation of the measured values of the elasticity of potassium vapors with an accuracy of up to 2%. P - pressure in ata; T - temperature in <sup>O</sup>K. There are 2 figures and 3 references: 1 Soviet.

Card 2/2

GRACHEV, N.V.; KOZHEVNIKOV, I.I.

Boundary layers of the Cretaceous and Paleogene systems of the northern Caspian Sea region. Trudy SCPK no.2:183-192 \*61.

(MIRA 14:11)

(Caspian Sea region--Paleontology, Stratigraphic)

### GRACHEV, P.I.

Reconditioning the teeth of standard levers of a ST-35 apparatus.

Vest.sviami 14 no.4:30 Ap \*54. (MLRA 7:6)

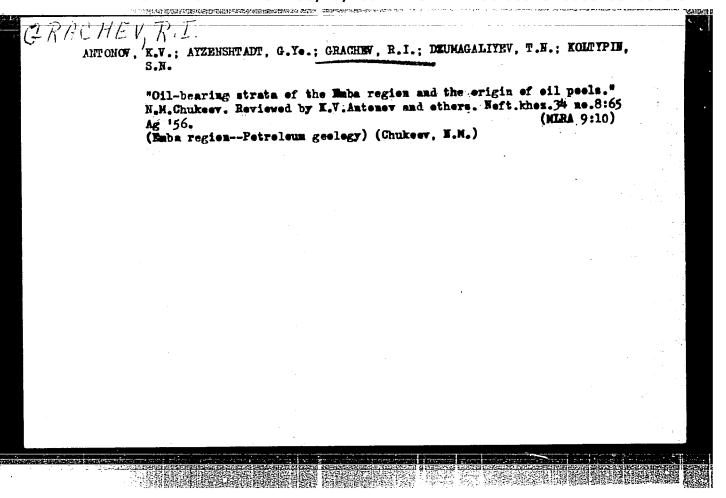
1. Glavnyy inchener Chelyabinskogo tsentral'nogo telegrafa. (Telegraph--Apparatus and supplies)

一次の行動性的性質を使用的性質的性質を使用的性質的性質的性質的性質的

GRACHEV, R.I.; DZHUMAGALIYEV, T.N.

Geological structure of the eastern part of the Ural-Volga interfluve.
Trudy Inet. nefti AH Kasakh. SSR no.1:5-16 '56. (MIRA 10:4)

(Ural Valley--Geology) (Volga Valley--Geology)



AUTHOR TITLE GRACHEV R.I., DONSKOVA G.M., RYGINA P.T. 20-2-49/67

New data on the stratigraphy and Distribution of calluvian and oxfordian deposits upon the territory of the near-Caspian depression.— (Novyye dannyye o stratigrafii i rasprostranenii otlosheniy kelloveya i oksforda na territorii Prikaspiyskoy

PERIODICAL ABSTRACT vpadiny,- Russian) Doklady Akademii Nauk SSSR 1957, Vol 113, Nr 2, pp 418-420 Un a large area of the Caspian depression below the marine layers of the lower Volga-deposit there lies an immense (400-600 m), even mass alternatingly consisting of loam, sand, limestone, and sometimes coal. Till recently any fauna was refused to this layer, except for single Lingula sp. and Pseudomomonotis sp. The plant-remains were determined as "middle jurassic" in a wider sense. For this reason this sediment mass was classified into the middle jurassic and the mainland: to the coastal some of the sea and the swampy mainland. By the investigation of these layers the authors obviously could ascertain in their series sediments of the shallow sea, especially of the Callovian and Oxfordian deposits. In the extreme Sotheast of the Emba-region there is on authentic middle jurassic a deposit of a rook layer mainly consisting of loam with thin intermediate layers of limestone, without any lime at the bottom, with a few plant-remains: higher up

CARD 1/5

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516510013-9"

20-2-49/57

New data on the stratigraphy and Distribution of callovian and oxfordian deposits upon the territory of the near-20-2-29/30 Caspian depression.

it becomes slightly ortaceous in parts. According to the fose sil determinations this mass doubtlessly have to be classified among the Callovian deposits. Its thickness is 110 m. Grey, sometimes slightly ordaceous, sandy loam with bhin intermediate layers of loamy marl and a fauna of the Oxfordian deposit, lies on it with a thickness of 30 m. In the Asnagularea below the lower Volga deposit in the upper part (formerly classified among the middle jurassic), sediments of all three stages of the mentioned deposits were discovered. They lay on authentic Bath-deposit sediments and are constructed alternatingly of limestone, loam with carbonised vegetabilic detritus, and slightly loamy marls. Lithologically they are identical with Bath and Callovy. Besides the fauna component mentioned by name, a characteristic pollen complex was found here, which does not reach down into the Bath and Bayos. Chiefly sackless pollen of the conventional-famyli Aggerells can be found here with two new kinds. The Callovian deposit is about 112 m thick. It is covered by the loam of the Oxfordian deposit (22 m thick) here, too. In the wewtern part of the depression, between the rivers Ural and Volga, several

CARD 2/5

New data on the stratigraphy and Distribution of callovian and oxfordian deposits upon the territory of the near-Caspian depression.

drillings in Aukitaychagyl (Novokasanka) were investigated. Here, too, the upper part of the cross-section (formerly regarded as middle jurassic) contains Callovian and Oxfordian sediments. On the Bath lies without noticeable interruption a packet of loam oretaceous in its upper part, 120 m thick, with a number of fossils, which are characteristic of the central and the lower Volga-region. Here the spore- and pollencomplex still appears, which is lacking further down and is mainly represented by new representatives of 6 conventional families. In the Oxfordian deposit here alternatingly sandoretaceous-loams with marls occur, which contain carbonised plant detritus and pyrite crystals. Foraminiferes typical for the Oxfordian depositi are found. They are 35 m thick. In the central part of the Emba-region the upper horizons under the lower Volga-deposit are represented by sediments from shallow water, which have no marine fauna. Spore- and pollencaplec is of typical Callovian kind. The Callovian and

CARD 3/5

New data on the stratigraphy and Distribution of callovian and oxfordian deposits upon the territory of the near-Caspian depression.

Oxfordian deposits can easily be sorted out.on diagrams by electrocarrottage. They are present in many summits of the region. Thus the existing of Callovian and Oxfordian deposits is demonstrated for the whole Embadistrict, except for the part of the salt-summits, which at that time had their most active period of development- However, at their peripheral parts of the summits they did not suffer any erosion. It can be maintained that the accumulation of sediments of this age occurred in the whole Caspian-depression. In a number of papers a new kind of plateau is shown in the central part of the Emba-region (Zhilokosinskoye, Tanatar- or Makatelevation). Its presence explains the occurrence of the Callovian and Oxfordian deposit in the most shallow facies, which are the nearest to the continental ones, without any marina fauna, however with a typical spore- and pollencomplex, which is clearly distinguished from that one of the Bath deposit. (8 citations from Slavic publications)

CARD 4/5

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516510013-9"

New data on the stratigraphy and Distribution of callovian and exfordian deposits upon the territory of the near-Caspian depression.

ASSOCIATION: Central Scientific Research Laboratory of the Kasakhstanneft'

(Kasakhstian-mineral oil)

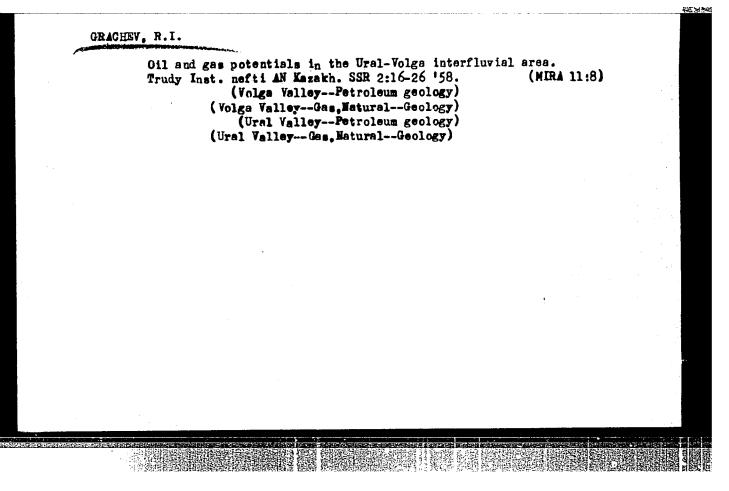
PRESENTED BY: N.M. STRAKHOV, Member of the Academy

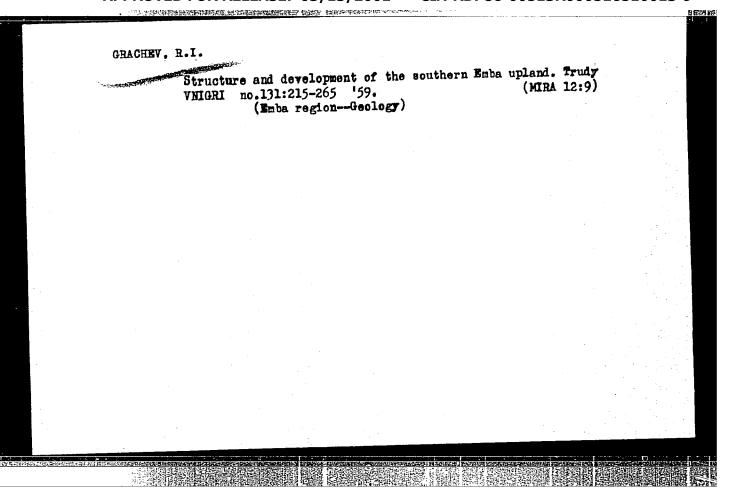
SUBMITTED: 20.1. 1956

AVAILABLE: Library of Congress.

CARD 5/5

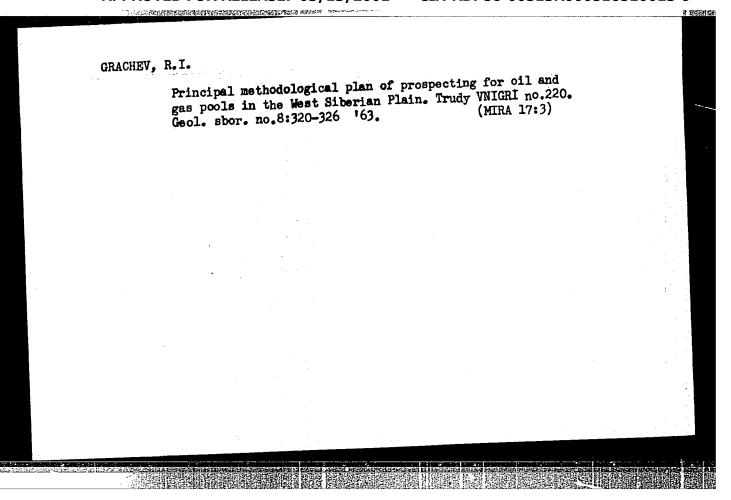
APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516510013-9"





IVANCHUK, P.K.; KOZLOV, I.G.; GRACHEV, R.I.

Geological results of exploratory drilling in the U.S.S.R. for the period 1947-1957. Trudy VNIGRI no.132:100-111 '59. (MIRA 17:1)



BOYARSKIKH, G.K.; GRACHEY, R.I.; MIRONOV, Yu.K.

Method of prospecting for possible zones of oil and gas accumulation in the West Siberian Plain. Trudy VNIGRI no.220. Geol. sbor. no.8:327-344 '63. (MIRA 17:3)

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516510013-9"

GRACHEV, Rostislav Ivanovich; BROYTMAN, Roman Yakovlevich; VFRESHCHAKO,
Igor' Aleksandrovich; ROZENEERG, Nikolay Mikhaylovich; IEYBSON,
M.G., nauchnyy red.; GINTSBURG, V.I., vedushchiy red.

[Determining the efficiency of geological prospecting; methodological instructions]. Opredelenie effektivnosti geologorazvedochnykh rabot; metodicheskie ukazaniia. Leningrad, Nedra, 1964. 84 p. (Leningrad. Vsesoiuznyi neftiano nauchno-issledovatel skii geologorazvedochnyi institut. Trudy, no. 229) (MIRA 17:6)

GRACHEV, R.I.; ANSIMOV, V.V.; BOYARSKIKH, G.K.; VERESHCHAKO, I.A.; MIN'KO, V.A.; MIRONOV, Yu.K.; SHIHIOV, V.G.; SHAMES, D.Z.; IONINA, I.H., vedushchiy red; CHOCHIA, N.G., red.

[Geological and economic efficiency in prospecting for oil and gas in the West Siberian Plain.] Geologo-ekonomicheskaia effektivnost¹ geologopoiskovykh i razvedochnykh rabot na neft¹ i gaz v Zapadno-Sibirskoi nizmennosti. Leningrad, Gostoptekhizdat, 1963. 199 p. map (insert. Leningrad. Vsesoiuznyi i neftianoi nauchro-issledovatel¹ skii geologorazvedochnyi institut. Trudy, no.206). (MIRA 17:10)

GRACHEV, S.

The role of trade-union organisations in perfecting the Soviet apparatus. Sov.profsoiusy 3 no.2:32-35 F 155. (MRA 8:4)

1. Predsedatel Moskovskogo obkom profsoyum rabotnikov gosuchrezhdeniy.

(Trade unions) (Russia-Politics and government)

GRACHEV, S., inzh.-podpolkovnik.

Dono in a tank repair workshop, Tankist no.3:50-51 Mr '58.

(Tanks (Military science)—Maintenance and repair) (MIBA 11:5)

(MIRA 15:12)

Closer to the economic work of financial organs. Fin. SSSR 23

no.11:75-77 N '62.

1. Predsedatel' Moskovskogo gorodskogo komiteta professional'nogo soyuza rabotnikov gosudarstvennykh uchrezhdeniy.

(Moscow-Finance) (Moscow-Auditing and inspection)

# The movement for the communist attitude toward labor is expanding. Fin.SSSR 37.no.3163-67 Mr '63. (MIRA 1614) 1. Predsedatel' Moskovskogo gorodskogo komiteta professional'nogo. soyuza rabotnikov gosudarstvennykh uchrezhdeniy. (Moscow---Financial employers) (Socialist competition)

GRACHEV, S. A., MURIN, A. N., NEFEDOV, V. P., ZAYZTSEV, V. M. (USSR)

"Use of Chemical Changes Accompanying Processes of Beta-Decay of RaE for the Synthesis of Organic Compounds of Polonium".

paper submitted for the Symposium on the Chemical Effects of Nuclear Transformation (IAEA) Prague, 24-27 Oct. 1960.

	V.D.; GRACHEV, S.A.	•	•	
	Paper chromatographic study of the chemical forms of RaE formed in the β-decay of RaD. Radiokhimiia 2 no.4:464-469 160.			
	(LeadIsotopes)	(BismuthIsotopes)	(MIRA 13:9)	
				* 14
				\$
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5.2500 5.3700(B) S/020/60/133/01/34/070 B011/B003

AUTHORS:

Murin, A. N., Nefedov, V. D., Zaytsev, V. M., Grachev, S.A.

TITLE:

Synthesis of Elemental-organic Compounds of Polonium by Using Chemical Changes Taking Place During the Processes

of Beta Decay of RaE

PERIODICAL:

Doklady Akademii nauk SSSR, 1960, Vol. 133, No. 1,

pp. 123 - 125

TEXT: The ability of the elements polonium, francium, and astatine to form elemental-organic compounds is a result of their position in the periodic system and of a general law discovered by D. I. Mendeleyev. This law was newly formulated by Academician A. N. Nesmeyanov (Refs. 1 and 2). The present paper describes the development of new methods of synthesizing the compounds mentioned in the title, which had been unknown so far. The method based on the utilization of chemical changes occurring during  $\beta$ -decay might be useful in this case (Refs. 7-11 for bismuth). The authors prove that the said polonium compounds (RaF) are formed by  $\beta$ -decay of RaE. RsE is a component of several aromatic

Card 1/4

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516510013-9"

Synthesis of Elemental-organic Compounds of S/020/60/133/01/34/070
Polonium by Using Chemical Changes Taking B011/B003
Place During the Processes of Beta Decay of RaE

CCl<sub>4</sub> (without treatment of the paper):  $R_f$ : TePh<sub>3</sub>Cl $\sim$ 0; TePh<sub>2</sub>Cl<sub>2</sub> 0.6 - 0.7; TePh<sub>2</sub> $\sim$ 1. Fig. 2 shows the distribution of the  $\alpha$ -activity among various chemical modifications of polonium on accumulation in Bi(RaE)Ph<sub>3</sub> crystals: PoPh<sub>2</sub>Cl<sub>2</sub> 15±6%; PoPh<sub>2</sub> 24±6%, and the sum of the remaining Po derivatives was 61±6%. Data are also given for CCl<sub>4</sub> and petroleum ether. Fig. 3 shows the results of chromatographing in ethyl acetate ( $R_f$  = 0.54). It may be seen that the chemical state has a strong effect on the yields of various RaF forms. This makes it possible to utilize chemical changes of various RaF forms. This makes it possible to utilize chemical changes occurring in  $\beta$ -decay for the synthesis of the Po compounds mentioned in the title. The authors thank G. A. Razuvayev, Corresponding Member of the AS USSR, and B. K. Preobrazhenskiy for their advice. There are figures and 15 references: 9 Soviet, 1 American, 4 German, and

Card 3/4

S/186/61/003/001/020/020
A051/A129

AUTHORS: Grachev, S.A., Mel'nikov, V.N., Ryukhin, Yu.A., Toropova, M.A.

PERTODICAL: Radiokhimiya, v 3, no 1, 1961, 116-118

TEXT: The radioactive isotope Cd 109 is formed when irradiating silver in a cyclotron according to the reaction: Ag 109(d, 2n) Cd 109. The Cd 109 decays cyclotron according to the reaction: Ag 109(d, 2n) Cd 109 the monochromatic by K-capture with a half-life of 470 days. The energy of the monochromatic by K-capture with a half-life of 470 days. The lang-lived isotope gamma-emission E = 87 kev. In addition to Cd 109 the long-lived isotope of silver Ag 110m (T=270 days) is also formed according to the reaction of silver Ag 110m (T=270 days) is also formed according to the reaction ag 109(d,p)Ag 110m. The problem of separating Cd 109 without a carrier is resilver and copper. Reference is made to certain other methods of Cd separasilver and copper. Reference is made to certain other methods of Cd separasilver and copper. Reference is made to certain other methods of Cd separasilver and copper. Reference is made to certain other methods of Cd separasilver and copper. Reference is made to certain other methods of Cd separasilver and copper. Reference is made to certain other methods of Cd separasilver and copper. Reference is made to certain other methods of Cd separasilver and copper. Reference is made to certain other methods of Cd separasilver and copper. Reference is made to certain other methods of Cd separasilver and copper. Reference is made to certain other methods of Cd separasilver and copper. Reference is made to certain other methods of Cd separasilver and copper. Reference is made to certain other methods of Cd separasilver and copper. Reference is made to certain other methods of Cd separasilver and copper. Reference is made to certain other methods of Cd separasilver and copper. Reference is made to certain other methods of Cd separasilver and copper. Reference is made to certain other methods of Cd separasilver and copper. Reference is made to

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516510013-9"

23885

S/186/61/003/001/020/020 A051/A129

Separation of Cd 109 without a carrier ...

carrier in the presence of very large quantities of copper and silver. The authors recommend the following method of Cd<sup>109</sup> separation without a carrier: The target on which the silver was placed was submerged into concentrated nitric acid and the silver layer was dissolved. After washing the target with distilled water, the combin ! solutions (nitric acid and aqueous) were transferred to a triple-mouth flowk (Fig 1). The solution was heated. The silver iodide and copper semi-iodide were precipitated by adding a 10% solution of HI, while mixing. The solution was separated from the precipitate through a porous quartzite filter with a pore size of 20-35 into a separating funnel. The precipitate was washed 2-3 times with a 1% solution of HI, after which the solution was poured into a quartzite container through the lower tap of the separating funnel and was evaporated until dry. The precipitate was processed twice with distilled concentrated HCl with subsequent evaporating until dry, and was dissolved in 10 ml of 2n HCl. The quent evaporating until dry, and was dissolved in 10 ml of 2n HCl. The further purification of Cd<sup>109</sup> from copper and traces of silver was conducted by using an ion-exchange column. The column with a diameter of 10 mm and a length of 70 mm was submerged in AB-17 (AB-17) resin with a grain size of 50-100 a. The resin was washed eliminating iron and transferred to a Cl-con-

Card 2/5

Separation of Cd 109 without a carrier ...

S/186/61/003/001/020/020 A051/A129

tainer. The initial solution was passed through a column at a rate of 0.15 ml per minute. Then the column was rinsed with 70-80 ml of 2 n HCl, whereby traces of copper were removed. Cd<sup>109</sup> was evaporated to the required volume. The radiochemical purity of the obtained sample was checked by taking the gamma-spectrum using a scintillation gamma-spectrometer with automatic registering of the gamma spectra. A NaI crystal was used as the scintillator, having a counting efficiency of the gamma quanta with a 100 kev energy equalling 18%. Fig 2 shows the gamma-spectrum of the initial nitric acid solution containing Cd<sup>109</sup> and Ag<sup>110m</sup>. The 87 kev energy peaks belong to Cd<sup>109</sup>, and the 447, 883, 936 and 1382 kev peaks belong to Ag<sup>110m</sup>. Fig 3 is the gamma-spectrum of the AgI precipitate. The gamma-spectrum of the Cd<sup>109</sup> sample formed without a carrier is shown in Fig 4. The presence of the only line with an energy of 87 kev in the spectrum indicates the radiochemical purity of the Cd<sup>109</sup> sample obtained. The integral change of the activity of all the gamma-lines with an energy over 100 kev proved that the radioactive contaminations of the sample are much below 0.1%. There are 3 graphs, 1 diagram and 7 references: 3 Soviet-bloc, 4 non-Soviet-bloc.

Card 3/5

23003

s/186/61/003/002/015/018 E111/E452

24.6720 (1482, 1563, 1138)

Nefedov, V.D., Vykhovtsev, V.L., Chi-Lan, Wu and

Grachev, S.A.

TITLE:

**AUTHORS:** 

Chemical changes occurring in β-decay of RaD which is part of the composition of radical-deficient

derivatives of lead

PERIODICAL: Radiokhimiya, 1961, Vol.3, No.2, pp.225-228

The authors note the complicated nature of changes occurring in the  $\beta$ -decay of the central atom of organic derivatives of lead. It was shown (Ref.1: V.D.Nefedov, V.I.Andreyev, ZhFKh, 31, 3, 563 (1957) and Ref.2: V.D.Nefedov, M.P.Bel'dy, ZhFKh, 31, 5, 986 (1957)) that with excess-radical and radical equivalent phenyl derivatives of RaD and ThB the bismuth isotopes produced on their decay can form a whole series of compounds. Their origin is complex and can be primary (due to rearrangement of the original molecule in  $\beta$ -decay processes not accompanied by internal conversion), secondary (due to recombination of fragments produced in  $\beta$ -decay processes accompanied by internal conversion) or tertiary (due to redistribution of daughter elements between compounds due to Card 1/4

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516510013-9"

23003

Chemical changes ...

S/186/61/003/002/015/018 E111/E452

chemical processes during analysis). It is difficult to establish the participation of adjacent molecules in some of the changes taking place. A possible way of solving this important problem is to study the chemical state of daughter atoms produced in the decomposition of radical-deficient lead derivatives (i.e. compounds whose molecules contain insufficient radicals to form the highest organic derivatives of bismuth, e.g. PbPh2Cl2). If appreciable quantities of such organic bismuth derivatives appear in the decomposition products, this would indicate that neighbouring molecules participate, and conversely. The problem is thus to compare the quantities of complete organic bismuth derivatives formed in the decomposition of, on the one hand, radical-equivalent and radical-surplus and, on the other hand, radical-deficient derivatives. The authors studied this with RaDPh, RaDPh3Cl and RaDPh2Cl2. They used the isotope-carriers method (described in Ref.l and 2) to study first the compounds of RaE formed during accumulation in crystals of RaDPh4 and RaDPh3Cl. The proportion of RaE existing as RaEPh; was determined by the precipitation and by the solvent-removal methods. In the first, the Pb(RaD)Ph3Cl or Card 2/4

23003 S/186/61/003/002/015/018 E111/E452

Chemical changes ...

Pb (RaD)Ph4 was dissolved in 15 ml of pyridine containing about 400 mg BiPh3 and 5 ml of BiPh3Cl2. The precipitate obtained on adding 30 ml of distilled water was filtered off and washed with water and then with two 10 ml portions of alcohol. After drying ether extraction was effected, the extract being filtered and evaporated. The Bi(RaE)Ph3 residue was purified by recrystallisation. In the solvent removal method, about 35 mg of Pb (RaD)Ph3Cl or Pb (RaD)Ph4 was dissolved as before, but the pyridine was removed by a stream of cold air; the residue was extracted with ether, the Bi(RaE)Ph3 obtained after removal of ether being purified as before. Similar procedure was used for RaE existing as RaEPh3Cl2. The two methods were also used for RaE compounds formed on accumulation in crystals of Pb(RaD)Ph2Cl2, but here warm pyridine was used. Special experiments showed that among the causes of discrepancies between the result of the two methods are tertiary changes. The results show that chemical changes in  $\beta$ -decay of the central atom (RaD) in element-organic derivatives of lead RaDPh4, RaDPh3Cl and RaDPh2Cl2 do not draw in surrounding molecules. Acknowledgments are expressed to A.N. Murin who made valuable suggestions on this work. There are Card 3/4

23003

Chemical changes ...

5/186/61/003/002/015/018 E111/E452

2 tables and 5 references: 3 Soviet-bloc and 2 non-Soviet-bloc. The reference to the English language publication reads as follows: R.R.Edwards, J.M.Day, R.E.Overman, J.Chem.Phys., 21, 9, 1555 (1953).

SUBMITTED: October 16, 1959

Card 4/4

NEFEDOV, V.D.; GRACHEV, S.A.; GRANT, Z.A.

Separation of p-tolyl derivatives of tellurium by paper chromatography. Zhur.ob.khim. 32 no.4:1179-1182 Ap '62. (MIRA 15:4)

1. Leningradskiy gosudarstvennyy universitet. (Tellurium—Analysis) (Paper chromatography)

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516510013-9"

MERCHANISM TO A STREET OF THE PROPERTY OF THE

S/079/63/033/001/001/023 D403/D307

AUTHORS:

Nefedov, V. D., Toropova, M. A., Grachev, S. A., and

Grant, Z. A.

TITLE:

Chemical changes during the B-disintegration of Rak

and their utilization for the synthesis of p-tolyl

derivatives of polonium

PERIODICAL: Zhurnal obshchey khimii, v. 33, no. 1, 1963, 15-18

TEXT: A discussion is first given of the possibility of using the chemical changes occurring during the disintegration of RaE compoundsm particularly aromatic organometallics, for the synthesis of Po derivatives, proposing that the nature of the parent molecule should exert a strong effect on the character of chemical changes during the β-disintegration. Confirmation of this proposition and application of the above chemical changes to synthesis of the organometallic compounds of Po, At, Fr, etc. were the chief objects of this work. The compounds studied were Bi(RaE)Tol<sub>3</sub> and Bi(RaE)-Tol<sub>3</sub>Cl<sub>2</sub>, where Tol=tolyl, using paper chromatography to separate Card 1/2

Chemical changes during ...

S/079/63/033/001/001/023 D403/D307

and identify the Po derivatives formed. Analogous Te compounds were used as carriers for the Po derivatives. The best eluents were found to be MeOH-H<sub>2</sub>O and (CH<sub>3</sub>)<sub>2</sub>CO-MeOH-H<sub>2</sub>O, taken respectively in the proportions of 3:1 and 4:3:2 by volume. The following new compounds were found: Po(p-Tol)<sub>2</sub>, (p-Tol)<sub>2</sub>PoCl<sub>2</sub>, (p-Tol)PoCl<sub>3</sub> and (p-Tol)<sub>3</sub>PoCl, as well as some inorganic Po. The relative proportions of these products confirmed that the nature and properties of the parent compound strongly affect the chemical changes. The advice and assistance of G. A. Razuvayev and A.N. Murin is acknowledged. There are 3 figures and 2 tables.

SUBMITTED: February 26, 1962

Card 2/2

S/079/63/033/002/001/009 D403/D307 Nefedov, V.D., Grachev, S.A. and Gluvka, S. AUTHORS: Study of chemical changes occurring during the TITLE: 6 - disintegration of RaE contained in its naphthyl derivatives Zhurnal obshchey khimii, v. 53, no. 2, 1963. PERIODICAL: 333 - 339 Continuation of earlier studies (DAN SSSR, 133, TEXT: 123 (1960); ZhOKh, 33, 15 (1963)) which showed that consequences of β - disintegration may be used to initiate hitherto unknown synthetic methods. Po was accumulated in  $(\alpha - C_{10}H_7)_3$  Bi(RaE) and  $(\alpha - C_{10}H_7)_3$ Bi(RaE)Cl2, which were irradiated with neutrons at a flux of 104 n/sec. cm2, and the products were separated by paper chromatography, using the corresponding Te compounds as carriers. The latter compounds were labelled with Te 127. Quantitative analysis of the chromatograms was carried out by the \$ - activity of Te 127 and by the d-activity of Po. The chromatographic procedure is described in some detail. (4-0,0H7), Po

Study of chem	nical changes	S/079/63/033/0 D403/D307	02/001/009	
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Mechanism of	chemical changes in	anic Po, $(2-6\%)$ , with a district state of $\beta$ - district states on the same states and the same states of	egration of ReE	
to singly cha	rged primary molecu	processes are those lar ions, the structu 4 figures and 5 table	re of which de-	
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MURIN, A.N.; NEFEDOV, V.D.; KIRIN, I.S.; CRACHEV, S.A.; GUSEV, Yu.K.; SAYKOV, Yu.P.

Formation of oxygen compounds of xenon during the -decay of 1911 in potassium periodate. Radiokhimiia 7 no.5:631-632 165.

(MIRA 18:10)

NEFEDOV, V.D.; ZHURAVLEV, V.Ye.; TOROFOVA, M.A.; GRACHEV, S.A.; LEVCHENKO, A.V.

Synthesis of some p-tolyl derivatives of polonium. Zhur. ob. khim. 35 no.8:1436-1440 Ag '65. (MIRA 18:8)

1. Leningradskiy gosudarstvennyy universitet.

MURIN, A.N.; KIRIN, I.S.; NEFEDOV, V.D.; GRACHEV, S.A.; GUSEV, Yu.K.

Chemical changes in the decay of lodine isotopes as a method of synthesizing xenon compounds. Dokl. AN SSSR 161 no.3:611-613 (MIRA 18:4)

Mr '65.

1. Fiziko-tekhnicheskiy institut im. A.F. Ioffe AN SSSR. Submitted September 21, 1964.

L 17371-66 EWT(m)/EWP(t) DIAAP/IJP(c) JI

ACC NR: AP6004509

SOURCE CODE: UR/0186/65/007/005/0631/0632

AUTHOR: Murin, A. N.; Nefedov, V. D.; Kirin, I. S.; Grachev, S. A.; Gusev, Yu. K.;

Saykov, Yu. P.

ORG: none

 $\mathcal{B}$ 

TITLE: Formation of oxygen-xench compounds during β-radiation of I<sup>133</sup> incorporated in potassium periodide 27 -7

SOURCE: Radiokhimiya, v. 7, no. 5, 1965, 631-632

TOPIC TAGS: xenon, oxide formation, beta radiation, iodine, radioisotope

ABSTRACT: Xenon oxides (XeO4 and XEO3) were prepared by  $\beta$ -radiation of potassium periodide containing radioactive  $J^{133}$  isotope according to the following scheme:

[133]O4]1- \$\int [133XeO4] \cdot \frac{133XeO3}{133XeO4}

The preparation procedure was as follows: helium gas was bubbled for 30 minutes at

UDC: 541.28 : 546.295

Card 1/2

L 17371-66

ACC NR: AP6004509

a rate of 26 ml/min through a solution of KJ<sup>133</sup>O<sub>4</sub>, and KJ<sup>133</sup> in 0.002 normal H<sub>2</sub>SO<sub>4</sub> to remove free xenon. The elemental iodine was removed from the gas stream by passing helium through a KOH-absorber. The xenon oxides were trapped on AG-5 activated carbon at liquid nitrogen temperature. The quantity of trapped xenon-133 was measured using an AI-100-1 analyzer. It was found that XeO<sub>4</sub> is unstable in acidic media and decomposes to XEO<sub>3</sub>. /Enitor's note: J is the Russian periodic symbol for iodine./

SUB CODE: 07/ SUBM DATE: OBJan65/ ORIG REF: 002/ OTH REF: 005

NEFEDOV, V.D.; GRACHEVA, L.M.; GRACHEV, State; PETFOV, L.N.

Chemical changes during beta-decay of RaE contained in p-phenethyl derivatives as a method of synthesis of analogous derivatives of polonium. Radiokhimiia 7 no.6:741-744, 165.

(MIRA 19:1)

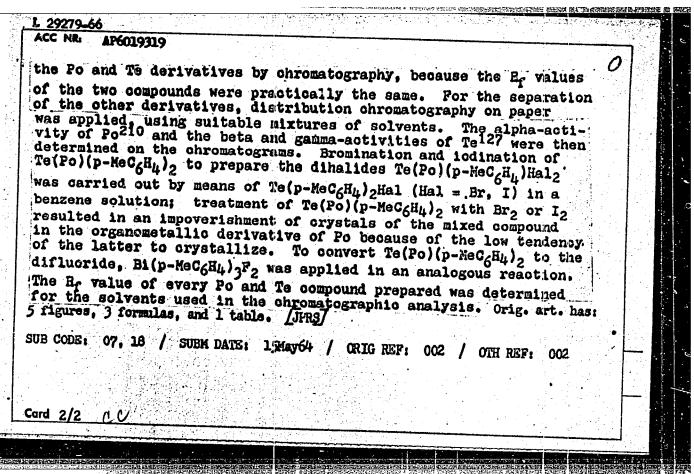
MURIN, A.N.; NEFEDOV, V.D.; KIRIN, I.S.; GRACHEV, S.A.; GUSEV, Yu.K.; SHAPKIN, G.N.

Beta decay of bromine isotopes as a possible method of synthesizing krypton compounds. Zhur.ob.khim. 35 no.12:2137-2140 D '65. (MIRA 19:1)

1. Fiziko-tekhnicheskiy institut imeni A.F. Ioffe AN SSSR. Submitted February 25, 1965.

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516510013-9"

SOURCE CODE: UR/0079/65/035/008/1436/1440 ENP(1)/EWT(m)/T L 29279-66 AP6019319 ACC NR: AUTHOR: Nefedov, V. D.; Zhuravlev, V. Ye.; Toropova, M. A.; Grachev, S. A.; Levchenko, A. V. ORG: Leningrad State University (Leningradskiy gosudarstvennyy universitet) TITLE: Synthesis of some p-tolyl derivatives of polonium SOURCE: Zhurnal obshchey khimii, v. 35, no. 8, 1965, 1436-1440 TOPIC TAGS: organic synthetic process, polonium compound, bismuth, tellurium, chemical precipitation, chromatography, bromination, iodinated organic compound, organometallic compound, radioisotope, radiation chemistry ABSTRACT: Po210 was separated from irradiated Bi by coprecipitation with Te from an HCl solution, using SnCl2. A mixture of TeCl4 and Pocly was then prepared by chlorination of elemental Te containing Te127 and Po210. Starting with Te(Po)Clu. p-tolyl derivatives of Po were prepared together with the analogous derivatives of Te by conventional chemical methods. Te(Po)(p-MeC6H4)2, the initial organoelemental compound from which Po(p-MeC6H4)2Hal2 (Hal = F, Cl, Br, I), Po(p-MeC<sub>6</sub>H<sub>4</sub>)<sub>3</sub>Hal(Hal = Cl, I), and Po(p-MeC6H4)3Cl.HgCl2 were prepared, could not be separated into



SVOBODA, Lyudvik [Svoboda, Ludvik], general armii; CRACHEV, S.I. [translator]; PETROV, F.P.[translator]; ARTEMOV, A.P., red.; SRIENIS, N.V., tekhnt red.

[From Buzuluk to Prague] Ot Buzuluka do Pragi. Moskva, Voenizdat, 1963. 405 p. Translated from the Ogeoh. (MIRA 16:6)

(Czechoslovakia---World War, 1939-1945)

CHUGAYEV, Yuriy Gennadiyevich; PLISKO, Valeriy Antonovich; EAVAROV, V.A.;

BOL'SHOV, V.M.; GRACHEV, S.N.; PASHKOV, A.A.; KACHKO, A.I.;

PLATONOV, S.A., polkovnik, red.; MEDNIKOVA, A.N., tekhm. red.

[Electronic digital computers | Bektronrye tsifrovye vychislitel'nye mashiny. Moskva, Voenizdat, 1962. 405 p. (MIRA 16:1)

(Electronic digital computers)

CHUGAYEV, Yuriy Gennadiyevich; PLISKO, Valeriy Antonovich; BAVAROV, S.F.;
BOL'SHOV, V.M.; GRACHEV, S.N.; PASHKOV, A.A.; KACHKO, A.I.;
PLATONOV, S.A., polkovnik, red.; MEDNIKOVA, A.N., tekhn. red.

[Electronic digital computers] Elektronnye tsifrowye vychislitel'nye mashiny. Moskva, Voenizdat, 1962. 405 p. (MIRA 16:2)

(Electronic digital computers).

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516510013-9"

SOV/124-58-10-11895

Translation from: Referativnyy zhurnal, Mekhanika, 1958, Nr 10, p 159 (USSR)

Zubov, V. Ya., Grachev, S. V. AUTHORS:

Relaxation Processes in a Steel Spring Strip (Relaksatsiya TITLE:

stal' noy pruzhinnoy lenty)

PERIODICAL: V sb.: Vopr. proyektir., izgotovleniya i sluzhby pruzhin.

Moscow-Leningrad, Mashgiz, 1956, pp 216-229

The effect of heat treatment on relaxation processes at temperatures ranging from 100 to 600°C was studied on a spring strip with ABSTRACT:

a cross section of 0.32x6.75 mm made of silicon steel EI142. Portions of the strip bent into a circular shape were inserted into steel rings of various diameters (the stresses in the strip did not exceed the elastic limit). After soaking at a certain temperature followed by a period of cooling, the strips were removed. The magnitude of "relieved" stresses was determined from the curvature of the strip. The intensity of stress reduction increased with increasing temperatures and increased with increasing mag-

nitude of the initial stress. At a temperature of 550°C, the

stresses were relieved completely by the mechanism of relaxation. Card 1/2

SOV/124-58-10-11895

Relaxation Processes in a Steel Spring Strip

Minimal relaxation stability was exhibited by quench-hardened steel. The greatest relaxation stability at temperatures ranging from 150 to 350° was observed in quenched steel which had been tempered at 450°. Alloying of spring silicon steel with Mo and W tends to increase the relaxation stability both at room temperature and at elevated temperatures.

M. Ya. Shashin

Card 2/2

KAGAN, Ya.A., kandidat tekhnicheskikh nauk; OSTROVSKIY, Ya.M., inzhener; GERZHOY, I.P., inzhener; GERZHOY, S.V. inzhener.

Improving the screw conveyor pulverized coal feeder and coal feeding unit for the MEI--Mosenergo system. Energomenhinostroenie 3 no.6:24-28 Je \*57. (MIRA 10:7)

(Boilers)

SOV/163-58-2-46/46

AUTHORS:

Zubov, V. Ya., Grachev, S. V., Grigor'yev, A. F.

TITLE:

The Influence of the Normal and the Isothermal Treatment on the Relaxation Stability of Spring Steel (Vliyaniye obychnoy i izotermicheskoy obrabotki na relaksatsionnuyu stoykost pruzhinnoy

stali)

PERIODICAL:

Nauchnyye doklady vysshey shkoly. Metallurgiya, 1958, Nr 2,

pp. 249-255 (USSR)

ABSTRACT:

The relaxation stability of metals and alloys is to a great extent determined by their atructure. In the present paper comparative investigations of the relaxation stability of spring steel of the types PET142 and USA under normal and isothermal treatment are described. The effect of the residual austenite on the relaxation process was discussed. The relaxation stability of spring steel treated the normal and the isothermal way depends on the conditions of relaxation. At low relaxation temperatures of the steel with martensite structure the relaxation stability is greater than in the case of a steel having a structure as in the complete decomposition of austenite. The change of the relaxation stability of the isothermally treated

Card 1/3

The Influence of the Normal and the Isothermal Treatment on the Relaxation Stability of Spring Steel

when the temperature of treatment is increased the relaxation stability first increases but then decreases again. Steel isothermally treated at high temperatures has the greatest relaxation stability at an increase in temperature. The residual austenite decreases the relaxation stability of the steel at low as well as at high relaxation temperatures, since at low temperatures a low resistance to plastic deformation exists, and at high temperatures a decomposition of the austenite takes place during the relaxation process. From this may be concluded that in the isothermal decomposition the presence of the residual austenite leads to a considerable decrease of the relaxation stability of the steel. There are 4 figures, 4 tables, and 9 references, 8 of which are Soviet.

ASSOCIATION:

Ural'skiy politekhnicheskiy institut (Ural Polytechnical Institute)

Card 2/3

GRACHEV, S. VI

AUTHORS: Zubov, V. Ya. and Grachev, S. V.

129-58-5-6/17

TITLE:

Resistance to Relaxation of Spring Strip at Room Temperature (Soprotivleniye relaksatsii pruzhinnov lenty pri

komnatnoy temperature)

PERIODICAL: Metallovedeniye i Obrabotka Metallov, 1958, Nr 5,

pp 20-23 (USSR)

ABSTRACT: A considerable number of instruments and mechanisms actuated by means of springs usually operate at temperatures approaching room temperature. The correctness of their readings and their reliability in operation depends mainly on the quality of the springs and one important property of the springs is their relaxation stability. A number of papers have been published on the relaxation stability of spring steels at elevated temperatures (Refs.1-4). In this paper the authors investigate the relaxation properties of spring strip of various grades of steel at room temperature under conditions of high bending stresses for which a method was used which was developed for investigating the relaxation stability of thin steel strip described in an earlier paper of one of the authors (Ref.5). The strip specimens are Card 1/3"charged" into rings, the diameter of which is so chosen

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129-58-5-6/17

Resistance to Relaxation of Spring Strip at Room Temperature

that the stresses do not exceed the bending strength of the material. There will be a gradual relaxation of the stresses of the strip in the rings. On removing the strip and determining the residual deformation as the direct elastic after effect it is possible to evaluate the relaxation by means of the following formula:

where h - the thickness of the strip;

 $\boldsymbol{\varrho}_{o}$  - the initial curvature radius of the strip; ep - curvature radius of the strip after relaxation.

The investigations were carried out using 0.32 x 6.75 mm specimens of the steels EI142, UlOA and an experimental spring steel EI722 of the following composition:
0.71% C, 2.49% Si, 0.51% Mn, 0.62% Cr, 0.6% W, 0.2% Mo,
0.016% P and 0.02% S. The conditions of heat treatment
are given. On the basis of the obtained results it is concluded that at room temperature the relaxation stability

Card 2/3 of the experimental steel, which has a higher Si content

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Resistance to Relaxation of Spring Strip at Room Temperature

and is additionally alloyed with tungsten and molybdenum, is considerably higher than for the steels EI142 and U10A, the latter having the lowest relaxation stability. Hardened, non-tempered, steel of the three investigated grades showed a reduced relaxation stability at room temperature. This is attributed to the partial process of decomposition of martensite as a result of the long duration effect of the stresses.

There are 2 figures and 6 Soviet references.

ASSOCIATION: Ural'skiy politekhnicheskiy institut im. S.M.Kirova (Ural Polytechnical Institute imeni S. M. Kirov)

AVAILABLE: Library of Congress.

Card 3/3

1. Springs-Stability-Test results

SOV/126-6-6-18/25

AUTHORS:

Zubov, V.Ya. and Grachev, S.V.

TITLE:

Relaxation Stability of Spring Steel as a Function of the Degree of Stability of the Structure (Relaksatsionnaya

stoykost' pruzhinnoy stali v zavisimosti ot stepeni

stabil'nosti struktury)

PERIODICAL:

Fizika Metallov i Metallovedeniye, 1958, Vol 6,

Nr 6, pp 1088 - 1094 (USSR)

ABSTRACT:

It was established that the relaxation stability of hardened and softened steel changes non-monotonically and that there is a certain interval of tempering temperatures for which the relaxation stability of the steel is at a maximum. The intensity of relaxation processes and the degree of development of such processes are influenced strongly by transformations in the investigated material. These transformations may be due to decomposition of the

solid solution, separation of one or another phase

component from the solid solution, coagulation, recrystal-lisation, etc. This aspect of the process of relaxation has not been adequately studied. Certain data can be found

from work relating to creep, the mechanism of which is very similar to that of relaxation but even there the

Cardl/4

Relaxation Stability of Spring Steel as a Function of the Degree of

available information is inadequate. The authors studied the influence of the degree of stability of the structure of spring steel strip on its relaxation stability under various temperature conditions. They investigated the influence of the temperature and the tempering time or the relaxation stability of spring steel. For the investigations, a standard silicon steel, EI142, was used and also an experimental 2.5% Si steel of the following composition: C 0.71%, Si 2.49%, Mn 0.51%, Cr 0.62%, Ni 1.17%, Mo C.20%, W 0.6%, S 0.02% and P 0.018%. Strips of both steels were first hardened. For obtaining differing degrees of structural stability, the hardened strip was tempered as various temperatures and heating durations. The influence of each of these factors on the relaxation stability was investigated separately. The specimens of both steels were tempered at 150, 250, 350, 450, 550 and 650°C for durations of 10 min. The elasticity limit of the tempered-strip specimens was tested according to a method described in earlier work of one of the authors (Ref 9). The test results are entered

Card2/4

Relaxation Stability of Spring Steel as a Function of the Degree of Stability of the Structure

in Table 1, p 1089. The results of the relaxation tests are described and entered in graphs, Figures 2-9. It was found that an increase in the stability of the structure does not necessarily lead to a monotonic increase in the relaxation stability. Depending on the conditions of relaxation and the structural state of the steel, an increase in the stability of the structure may bring acout a drop in the resistance of the material against relaxation. The role of the slip mechanism of relaxation will be the greater the higher the structural stability of the steel and the higher the initial stresses. The validity of a general relation can be discerned as regards the characteristic of the relaxation curves of two differing grades of spring steel. The relaxation stability of the experimental steel (denoted by EI722 in the paper) is higher than the relaxation stability of the steel EI142 for all the investigated preliminary tempering temperatures and relaxation temperatures.

Card3/4

SOV/126-6-6-18/25

Relaxation Stability of Spring Steel as a Function of the Degree of Stability of the Structure

> There are 9 figures, 2 tables and 9 references, all of which are Soviet.

ASSOCIATION:

Ural'skiy politekhnicheskiy institut imeni S.M.Kircva (Urals Polytechnical Institute imeni S.M. Kirov)

SUBMITTED:

July 6, 1956

Card 4/4

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516510013-9"

Plastic and elastic aftereffect in spring bands. Ixv.vys.ucheb.
EAV.; ohern.met. no.3:59-64 '60. (MIRA 13:4)

1. Ural'skiy politekhnicheskiy institut.
(Springs(Mechanism)) (Strains and stresses)

ZUBOV, V.Ya.; GRACHEV, S.V.; FESIN, Ya.A.

Stress relaxation during martensite transformations of residual austenite. Izv.vys.ucheb.mav.; chern.met. no.5:115-120
160. (MIRA 13:6)

1. Ural'skiy politekhnicheskiy institut.
(Steel--Metallography) (Strains and stresses)

ZUEOF, V.Ya., doktor tekhn.nauk prof.; GRACHEV, S.V., inzh.

Improving the quality of steel spring strips. Stal' 20 no.9:849-851
S '60. (MIRA 13:9)

1. Ural'skiy politekhnicheskiy institut.
(Steel--Heat treatment) (Springs (Mechanism)--Testing)

ZUBOV, V. Ya.; GRACHEV, S.V.; TSEYTLIN, A.M.

Stress relaxation during the tempering of high-speed steel.

Fiz. met. metallowed 11 no.3:465-466 Mr '61. (MIRA 14:3)

1. Ural'skiy politekhnicheskiy institut im. S. M. Kirova.

(Tool steel—Heat treatment)

(Strains and stresses)

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27929 s/133/61/000/006/014/017 A054/A129

AUTHORS:

Zubov, V. Ya., Doctor of Technical Sciences, Sokolov, N. V., Candidate of Technical Sciences, Krasil nikov, L. A., Grachev, S. V., Engineers

TITLE:

Deformation of metastable austenite and strength of steel strip

PERTODICAL: Stal

Stal Ano. 6, 1961, 549-551

TEXT: As a result of extensive research new ways were found to increase the strength of steel. Based on P. P. Anosov's studies, V. D. Sadovskiy (Ref.2: L. V. Smirnov, Ye. N. Sokolkov and V. D. Sadovskiy: Proceedings of the Institute of the Physics of Metals UFAN, 1956, no. 18, 35-36) put forward the suggestion that the excellent mechanical properties of Damascus blades were due to a combination of forging and hardening. With this theory in mind and the knowledge that the strength of alloyed steels could be raised by plastic deformation of austenite in supercooled condition, a so-called "thermo-mechanical" treatment was established for 65 (650) and 31/142 (EI142) type 2-mm thick spring wires. In the tests the wire was deformed (flattened) after heating until austenite formation and after cooling in tin bath from 380-400°C (Fig. 2). The strip (0.7 x 2.63 mm) processed by the new method in the laboratory was annealed at various

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Deformation of metastable austenite ...

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temperatures. Next the strength limit, the quantity of residual austenite and the microstructure of the strips under hardened condition were examined. It was found that by tempering at a temperature of up to 350°C with a holding time of 5 minutes the strength limit of EI142 steel increased to 300 kg/mm<sup>2</sup>. Maximum strength for 65G steel (280 kg/mm<sup>2</sup>) was obtained at a lower annealing temperature (300°C, holding time: 5 minutes). The transformation of austenite in 65C steel during annealing takes place more quickly than in EI142 steel. At 300°C and a holding time of 5 minutes the amount of residual austenite is no more than 10% in 65G steel, while at 360°C and a holding time of 1 minute nearly the entire quantity of austenite will be transformed. The microstructure of the test steels after flattening (with supercooled austenite and upon cooling at room temperature) displays elongated, dark grains with curved sliding surfaces in dense arrangement. These are evidently the products of the second stage of austenite transformation, which develops under the effect of plastic deformation on the disintegration of supercooled austenite. The tests were carried out with the cooperation of Engineer Yu. P. Surkov and Technician A. C. Lysenko. There are 5 figures and 7 references: 4 Soviet-bloc and 3 non-Soviet-bloc. ASSOCIATION: Beloretskiy staleprovolochnyy zavod (Beloretsk Steel-Wire Flant) Ural skiy politekhnicheskiy institut (Ural Polytechnical Institute)

Card 2/3

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5/129/62/000/001/004/011
                                          E073/E483
Zubov, V.Ya., Doctor of Technical Sciences,
            Grachev, S.V., Surkov, Yu.P., Engineers
           1454, 1045
             Influence of thermomechanical treatment on the
 1.1700
 PERIODICAL: Metallovedeniye i termicheskaya obrabotka metallov,
AUTHORS &
          The authors studied the possibility of using thermo-
TITLE :
 mechanical treatment in the drawing of carbon and low-alloy steel
  wire (L.A. Krasil'nikov and A.G. Lysenko participated in the
                  The chemical compositions (%) of the steels
  investigated are given as follows:
                                                                 0,006
  experiments).
                                                          0.019
                                             0.10
                                                                  0.023
                                      90.08
                                                    0.20
                                                          0.019
                                             0.12
                                                                  0.017
                                       0.12
                                                    0.10
                                                          0.009
                                             0.20
                                0.18
                   0.71
                                       0.09
                          0.20
                                                         The wire was
   77 (U7A)
    The initial wire diameters were 1.95 and 2 mm. The wire was heated for the drawing operation to 920 to 940°C by passing an
                                              0.10
                   1.01
                                 0.23
   710A (U10A)
    The initial wire diameters were 1.95 and 2 mm.
    65 (65G)
   Э́Й 142 (EI142) 0.66
     card 1/3
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33461 \$/129/62/000/001/004/011 E073/E483

Influence of thermomechanical ...

electric current through it and cooled in a lead bath to 320 - 350 °C. The speed of movement of the wire was 10 m/min. The wire was deformed in a single pass (short incubation period) by 5 to 32%, using a soap-graphite lubricant. The final After this, thermomechanical cooling after drawing was in air. treatment specimens of the wire were tempered under laboratory conditions at 100 to 500°C with a holding time of 1.5 min. strength of the wire drawn whilst the austenite was in the super-cooled state was very high. It was highest for the steel EI142, i.e. 306 kg/mm<sup>2</sup> (32% reduction and tempering at 350°C for Further experiments were carried out exclusively on this The hardness after thermomechanical treatment was higher by about 4 HRC units than for the same material quenched in In addition, hardness of thermomerhanicallythe ordinary way. treated material decreased more slowly with increasing tempering temperature than that of the same material after step-wise These differences were attributed to smaller grainquenching. size of martensite and presence of strain-hardened Even after austenite in thermomechanically treated steel. Card 2/3

33461

Influence of thermomechanical ...

S/129/62/000/001/004/011 E073/E483

3 minutes tempering at relatively high temperatures (450 to 500°C), hardness of thermomechanically treated steel was practically equal to that of the same material after the conventional hardening treatment. The strength of the wire increased with increasing reduction but there was a drop in strength after reductions not exceeding 6 to 8%. If reductions of the order of 30% are used (followed by tempering for 1 min at temperatures not exceeding with a strength of the order of 300 kg/mm<sup>2</sup>. There are 5 figures, 1 table and 1 Soviet-bloc reference.

ASSOCIATION: Ural skiy politekhnicheskiy institut
(Ural Polytechnical Institute)

Card 3/3

ZUBOV, V.Ya., doktor tekhn.nauk, prof.; GRACHEV, S.V., inzh.

Effect of preliminary plastic deformation on the properties of spring bands following heat treatment. Metalloved.i term.obr.met. no.2:46-47 F '62. (MIRA 15:3)

1. Ural'skiy politekhnicheskiy institut. (Steel—Heat treatment)

ZUBOV, V. Ya.; GRACHEV, S. V.

Phenomena of anomalous stress relaxation in beryllium bronzu.

Fiz. met. i metalloved. 14 no.4:602-607 0 '62.

(MIRA 15:10)

1. Ural'skiy politekhnicheskiy institut imeni S. M. Kirova.

(Beryllium bronze—Testing)

(Strains and stresses)

1 2011 C-65 FMT (m) /FMA (d) /T/BAP (t) /BMP (b) ASD (m)-3 MJW/JD/JXT (CZ)

ACCESSION NR: AR4044542

S/0277/64/000/006/0012/0012

SOURCE: Ref. zh. Mashinostr. mat., konstr. i raschet detal. mash. Otd. vy\*p, Abs. 6.48.77

AUTHOR: Grachev, S. V., Zubov, V. Ya.

L.

TITLE: Stress relaxation in austenitic steels

CITED SOURCE: Sb. Relaksats. yavleniya v met. i splavakh. M., Metallurgizdat, 1963, 309-312

TOPIC TAGS: austenitic steel, relaxational stability, stress relaxation, tempering temperature, structural metastability, steel mechanical property/steel 4Kh18N10S3, steel 4Kh18N15S2V2M, steel 2Kh18N9, steel E1481

TRANSTATION: The study concerned the relaxational stability (RS) of austenitic steels 4Kh15N10S3 and 4Kh16N15S2V2M ( & = 50 kg/mm<sup>2</sup>) when subjected to plastic deformation and tempored for 1 hr. in the 200-600C range. Stress relaxation (SR) tests were carried out on strip (0.36 x 4.5 mm cross-section) or wire (diam. = v.o mm) samples. Change in E-value during the heating process, predetermined dynamically, was considered when calculating stresses. The dependence of the extent of SR was plotted against the tempering Card 1/2

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ACCESSION NR: AR4044542

temperature (TT) at which the steel was preliminarily treated. The RS of steel increased as TT mass to levels of 400 - 500C. A similar TT corresponds to the peak tensile RS.

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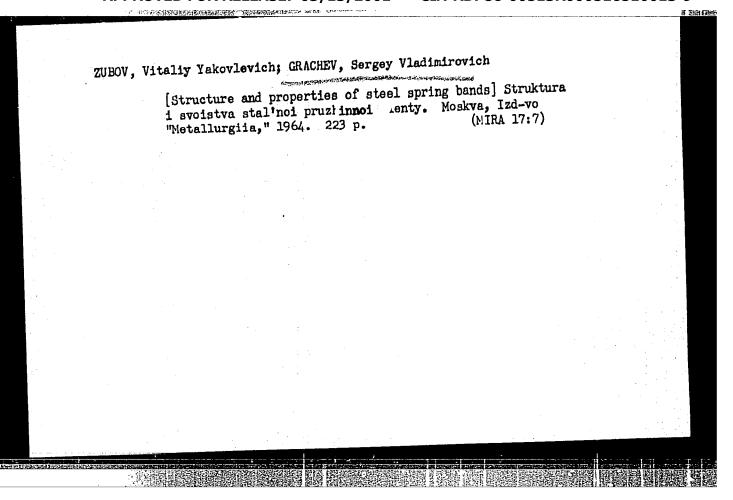
The related R

GRACHEV, S.V.; ZUBOV, V.Ya.

Stress relaxation in hardened steel in the first stage of tempering. Fiz. met. i metalloved. 15 no.6:854-859 Je '63. (MIRA 16:7)

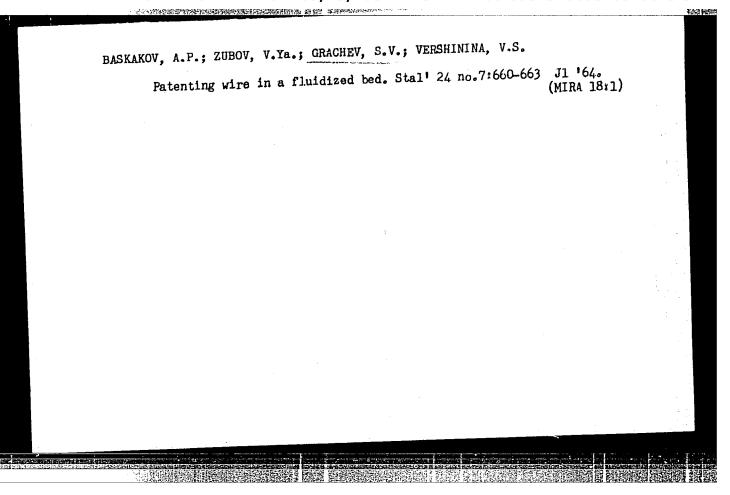
1. Ural'skiy politekhnicheskiy institut imeni Kirova.
(Steel—Heat treatment)
(Strains and stresses)

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000516510013-9"



== (x)/e;p(w)/epe(t)-2/eva(d)/I/evp(t)/evp(b) 5 5.6 AF5002 FF . W AUTHOR: Grachev, S. V.; Zubov, V. Ya. TITLE: Effect of structure stability on stress relaxation in alloys SOURCE: Fizika metallov i metallovedeniye, v. 18, no. 6, 1964, 909-914 TOPIC TAGS: stress relief mechanism, USA alloy, Ellic alloy, El722 alloy, communication plastic deformation inetastable structure, shear deforma-ABSTRACT: The mechanism of stress relief in alloys with metastable structures and examined. Stress, relief-temper temperature of the were constructed for That and E1772 steels subjected to temperate from the moves at 100-606C The position of the minima detended on the relaxaor grounder and type of steel. At expense of shearing components of plastic deformation was developed for the appearance of The general level of relief redictance was 100-150 degrees higher and with  $M_0 = W = Cr$  and St. thum a contract of the contract while the Card 1/2

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ZUBOV, V.Ya.; BASKAKOV, A.P.; GRACHEV, S.V.; ZAVAROV, A.S.; MALIKOV, G.K.

Characteristics of wire patenting in a fluidized bed. Izv.
vys. ucheb. zav.; chein. met. 8 no.10:116-119 '65.

(MIRA 18:9)

1. Ural'skiy politekhnicheskiy institut.

ZUBOV, V.Ya.; GRACHEV, S.V.; HYBAKOVA, M.F.; KIR'YANOVA, N.P.

"Hereditary" properties of thermomechanically treated steel.

Fiz. met. i metalloved. 20 no.3:424-427 S \*\*65.

(MIRA 18:11)

1. Ural'skiy politekhnicheskiy institut imeni S.M.Kirova.

ZUBOV, V.Ya.; BASKAKOV, A.P.; GRACHEV, S.V.; MALIKOV, G.K.; ZAVAROV, A.S.

Patenting in a fluidized bed with pittot plant equipment. Stal' 25 no.7.664-665 Jl '65.

1. Ural'skiy politekhnicheskiy institut.

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	L 44401-66 EWT(m)/EWP(w)/T/EWP(t)/ETI IJP(c) JD  ACC NR. AP6023640 SOURCE CODE: UR/0149/66/000/002/0129/0134  AUTHOR: Zubov, V. Ya.; Grachev, S. V.; Kirillov, Yu. L.; Spiridonova, L. M.; Norkina,  E. B. O/  ORG: Department of Metallurgy, Ural Polytechnic Institute (Kafedra metallovedeniya Ural'skiy politekhnicheskiy institut)  TITLE: Study of mechanical properties and relaxation stability of Cu-Ti alloys  SOURCE: IVUZ. Tsvetnaya metallurgiya, no. 2, 1966, 129-134  TOPIC TAGS: copper containing alloy, titanium containing alloy, chromium containing alloy, mechanical property, tensile strength, elastic modulus, stress relaxation, temperature dependence  ABSTRACT: The effect of certain factors of stress relaxation and other mechanical properties of five Cu-Ti alloys was studied. The alloys had Ti contents ranging from 1.10 perties of five Cu-Ti alloys was studied. The alloys had Ti contents ranging from 1.10 perties of five Gu-Ti alloys had Cr contents of 0.52 and 1.00%. After vacuum melting to 5.50%; two of the alloys had Cr contents of 0.52 and 1.00%. After vacuum melting to 5.50%; two of the alloys had Cr contents of 0.52 and 1.00%. After vacuum melting to 5.50%; two of the alloys had Cr contents of 0.52 and 1.00%. After vacuum melting to 5.50%; two of the alloys had Cr contents of 0.52 and 1.00%. The best of 0.50 and 500°C for 1 hr, quenched into water and cold worked 20, 40, 60 and which was heated to 860°C for 1 hr, quenched into water and cold worked 20, 40, 60 and which was heated to 860°C for 1 hr, quenched into water and cold worked 20, 40, 60 and 80%. Aging was carried out at 300, 350, 400, 450 and 500°C for 1 to 5 hrs. The best strengths were obtained by aging at optimal temperatures for 2 hrs. Tensile strengths strengths were obtained by aging at optimal temperature for all alloys in	· · · · · · · · · · · · · · · · · · ·
	80%. Aging was carried out aging at optimal temperatures for 2 hrs. identified strengths were obtained by aging at optimal temperature for all alloys in and relative elongations are given as functions of aging temperature for all alloys in and relative elongations are given as functions of aging temperature for all alloys in UDC: 669.35'295:669.018.2	
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the quenched and quenched + 60 % deformed condition. The highest strength (118 kg/mm?) was for 5.50% Ti + 0.52% Cr additions. Elastic moduli and electroconductivities for all alloys after quenching + 40% deformation are given. Maximum moduli were observed at aging temperatures of 400-450°C and for all alloys the limit was 60-70 kg/mm<sup>2</sup>. The best heat treatment, resulting in optimum combinations of strength and ductility, was obtained after deforming the quenched alloys 40-60% and aging at 400-450°C. These properties are considered to be ideal for replacing Cu-Be alloys used in springs. Alloying of Cu with Ti and Cr increased the electrical conductivity after quenching, however, this dropped considerably upon aging as a result of second phase decomposition. The elastic modulus, determined by the dynamic method, is given as a function of aging temperature and compared with beryllium bronze BrB2. This modulus rose sharpest for BrB2 indicating a faster decomposition of the solid solution. By increasing the managed of the solid solution. ing the Ti content the dynamic modulus decreased, probably as a result of a lowered interatomic bonding. Relaxation tests (relative relaxation stability as a function of time) were run at 200 and 400°C and the results were compared to BrB2. The Cu-Ti alloys had 4-15 times the relaxation stability at 400°C of BrB2. Again the best alloy was the 5.50% Ti + 0.52% Cr. Orig. art. has: 5 figures, 1 table, 1 formula.

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	AGG NR: AP6013359 (W)/T/ENP(t)/ETI IJP(e) JD SOURCE CODE: UR/0370/66/000/002/0076/0084
	AUTHOR: Zubov, V. Ya. (Sverdlovsk); Baskakov, A. P. (Sverdlovsk); Grachev, S. V. (Sverdlovsk); Zavarov, A. S. (Sverdlovsk); Antifeyev, V. A. (Sverdlovsk)
	TITLE: Patenting of wire in a fluidized bed
1	TITLE: Patenting or wire in a
	SOURCE: AN SSSR. Izvestiya. Metally, no. 2, 1966, 76-84
	ABSTRACT: The possibility of constructing an integrated unit for patenting wire in which the heating and cooling of the wire are carried out in a fluidized bed of fine-which the heating and cooling of the wire are carried out in a fluidized bed of fine-which the heating and cooling of the wire are carried out in a fluidized bed of fine-which the heating was studied on specimens of UTA, USA, U9A, and EI-142 steels. The grained material was studied on specimens of u7A, USA, U9A, and EI-142 steels. The grained material was studied on specimens of u8A, U9A, and EI-142 steels. The grained material was studied on specimens of u8A, U9A, and EI-142 steels. The grained material was studied bed was rate to correspondingly reduce the length of by a factor of up to 6, or at the same rate to correspondingly reduce the length of the heating systems as compared to the existing fuel-oil and electric furnaces. By the heating gas in a fluidized bed where oxygen is deficient, a nonoxidizing atmosphere burning gas in a fluidized bed where oxygen is deficient, a nonoxidizing atmosphere burning gas in a fluidized bed where oxygen is deficient, a nonoxidizing atmosphere burning gas in a fluidized bed where oxygen is deficient, a nonoxidizing atmosphere burning gas in a fluidized bed where oxygen is deficient, a nonoxidizing atmosphere burning gas in a fluidized bed where oxygen is deficient, a nonoxidizing atmosphere burning gas in a fluidized bed where oxygen is deficient, a nonoxidizing atmosphere burning gas in a fluidized bed where oxygen is deficient, a nonoxidizing atmosphere burning gas in a fluidized bed where oxygen is deficient, a nonoxidizing atmosphere burning gas in a fluidized bed where oxygen is deficient, a nonoxidizing atmosphere burning gas in a fluidized bed where oxygen is deficient, a nonoxidizing atmosphere burning gas in a fluidized bed where oxygen is deficient, a nonoxidizing atmosphere burning gas in a fluidized bed where oxygen is deficient, a nonoxidizing atmosphere burning gas in a fluidized bed where oxy
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元以位置其近距离的1980年64年12年12月12日20年2月12日 1842 2月2月2日 JD/HW IJP(c) ENT(m)/ENP(w)/T/EWP(k)/ENP(t)/ETI SOURCE CODE: UR/0126/65/020/003/0424/0427 110 L 46284-66 ACC NR: AP5025328 38 AUTHOR: Zubov, V. Ya.; Grachev, S. V.; Rybakova, M. F.; Kir'yanova, N. P. B ORG: Ural Polytechnic Institute im. S. M. Kirov (Ural'skiy politekhnicheskiy institut) TITLE: Problems of "heredity" of properties from thermomechanical treating of steel SOURCE: Fizika metallov i metallovedeniye, v. 20, no. 3, 1965, 424-427 TOPIC TAGS: mechanical heat treatment, spring steel, metal property, annealing, tempering, durability, elasticity, hardness, toughness ABSTRACT: The effect of additional tempering and annealing on thermomechanically treated, samples of spring strip was studied and the secondary treatment was shown to eleminate the favorable effects of the thermomechanical process. Samples of 0.4 x 4mm EI142 and U7A steel strip were austenized at 900C, precooled at 320C, rolled, and additionally sustenized at 860C and 7.2 m/min rate for approximately 30 sec, or at 860C in oil. Prior to the additional tempering some samples were annealed at 450-550C or at 300C. All samples, either after the primary thermomechanical treatment or after the additional heat treatment, were annealed 1-5 min at 200-500C and tested for strength, elasticity, toughness, and hardness. Samples, which had been tempered and annealed, but not mechanically treated, were similarly tested. UDC: 669.14.018.295 1/2 Card

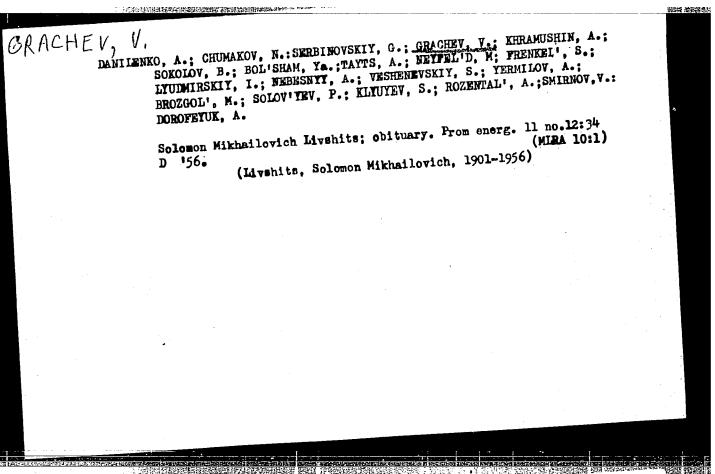
dditional heat treating decreased the mechanic ustenite to the level of strips obtained by ordinavorable properties occurs in the additional terroperties are preserved, causing an increase of figures and 1 table.						•
orig. art. has: 4 figures and 1 table.  SUB CODE: 11/ SUBM DATE: 24Sep64	/	ORIG REF:				
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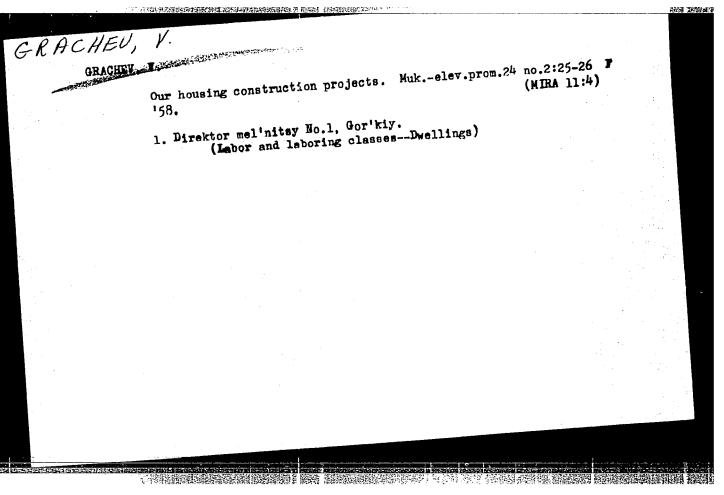
THE RESIDENCE OF THE PROPERTY OF THE PARTY O ORACHEV, V., (Novo-Ladoshskiy Rayon, Leningradskya Oblast) Dec 53 USSR/Electronics - Radiofication "Three Letters on Progress of the Radiofication Plan, "A. Vesset (Velikiye Luki), V. Grachev (Novo-Ladozhskiy Rayon, Leningradskaya Oblast), P. Lunev (Sevastopol') Radio, No 12, pp 8-9 Vesset discusses progress of radiofication in the Oktyabr'skiy and other rayons, and work of Fedulov, remarks that low power radio centers of the Min of Comm, esp the TUB-100, are overloaded and do not 276132 provide good service. Grachev blames poor planning for slow radiofication in Novo-Ladozhskiy Rayon. Lunev suggests that publications for radio amateurs lay more stress on radiofication problems.

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CIA-RDP86-00513R000516510013-9



GRACHEV, V.: SAFRONOV, A. (g.Bronnitsy)

Simplest intercommunication unit. Radio no.2:52 F 160. (MIRA 13:5)

(Intercommunication systems)

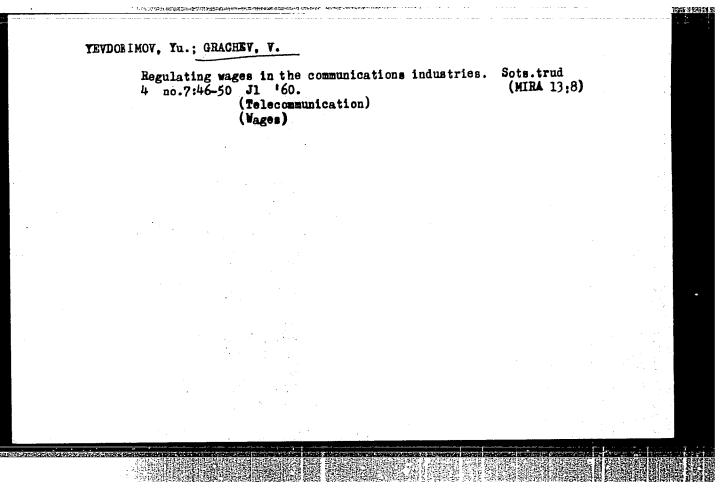
ALEXSEYEV, N.; GRACHEV, V.; MAILYEVA, A.; MENZHINSKIY, G.; NOVOZHILOV, V.;

SHARAGIV, A.; URVICHEV, P.

Over-all mechanization and electrification of the production.

Vop. ekon. no.3:100-110 Mr '60. (MIRA 13:2)

(Ehomutovka District--Farm mechanization) (Rural electrification)



	Compiling technological charts on a collective flax for sel'khoz. 20 no.6:19-23 Je '60.	arm. (MIRA	Tekh. 13:10)	:
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